



An entry to 1,6-dioxaspiro[4.6]undecanes and 1,7-dioxaspiro[5.6]dodecanes

Anthony Ollivier^a, Marie-Eve Sinibaldi^{a,b,*}, Loïc Toupet^c, Mounir Traïkia^{a,b}, Isabelle Canet^{a,b,*}

^a Clermont Université, Université Blaise Pascal, Laboratoire SEESIB, BP 10448, F-63000 Clermont-ferrand, France

^b CNRS, UMR 6504, SEESIB, F-63177 Aubiere, France

^c Groupe Matière Condensée et Matériaux, Université Rennes 1, Avenue du Général Leclerc, F-35042 Rennes, France

ARTICLE INFO

Article history:

Received 29 April 2010

Revised 31 May 2010

Accepted 1 June 2010

Available online 8 June 2010

Keywords:

Spiroketal
Cyclization
Hydrazone
Alkylation

ABSTRACT

Ketones **11a–c** obtained by iterative alkylation of acetone *N,N*-dimethylhydrazone with iodides **6** and **8a,b** or epoxide **9** followed by SiO₂/H₂O-induced cleavage of the hydrazone were quantitatively transformed into 1,6-dioxaspiro[4.6]undecanes **12a,c** and 1,7-dioxaspiro[5.6]dodecanes **12b** using Yb(OTf)₃ in CH₃CN.

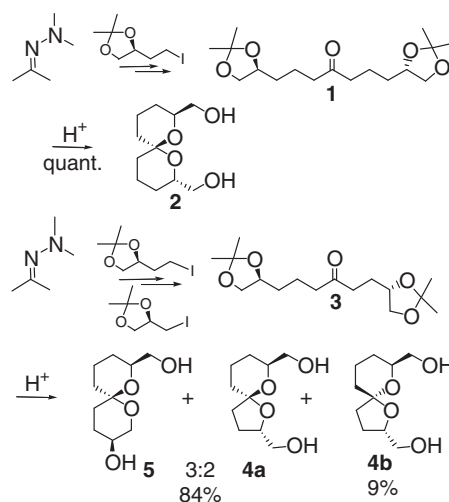
© 2010 Elsevier Ltd. All rights reserved.

Spiroketal form the common structural part of many biologically active, naturally occurring products issued from various sources including insects, plants, fungi, and marine organisms.¹ The mainly encountered structural variations relate to the size of the two spiranic cycles (spiro[4.4], spiro[4.5], and spiro[5.5] systems) and the presence or not of alkyl or alkoxy substituents on the spirocyclic skeleton. Because of their wide range of biological activities, spiroketals have seen a sustained interest in the development of methods for their preparation—mainly for 1,7-spiro[5.5]undecane and 1,6-spiro[4.5]decane systems—and numerous strategies have been reported and reviewed.² An interesting but rarer structural variation is exhibited by seven-membered ring spiroketals which could be formed stereoselectively but often in moderate yields.³

We investigated these last years' short routes to spiroketals and analogues. Our general strategy led to the elaboration of linear key-ketones of type **1–3**, obtained by iterative substitution of conveniently protected acetone,⁴ followed by a one-step deprotection—spiroketalization cascade. Thus, in spiro[5.5] series, we described the efficient synthesis of dihydroxymethyl-substituted 1,7-dioxaspiro[5.5]undecane **2**⁵ and demonstrated the facility in incorporating supplementary heteroatoms in the two cycles, starting from acetone *O*-benzyloxime.⁶ In spiro[4.5] series, we evaluated our strategy through the elaboration of 1,6-dioxaspiro[4.5]decane.⁵ Exposure of ketone intermediate **3** to acidic conditions led to a mixture of the targeted [4.5]-spiroketal, as the anomeric **4a** and

non-anomeric **4b** forms, but and contrarily to spiro[5.5] series, accompanied with the 1,7-dioxaspiro[5.5]undecane **5**—resulting from a double attack of primary alcohol functions on the keto group (Scheme 1).

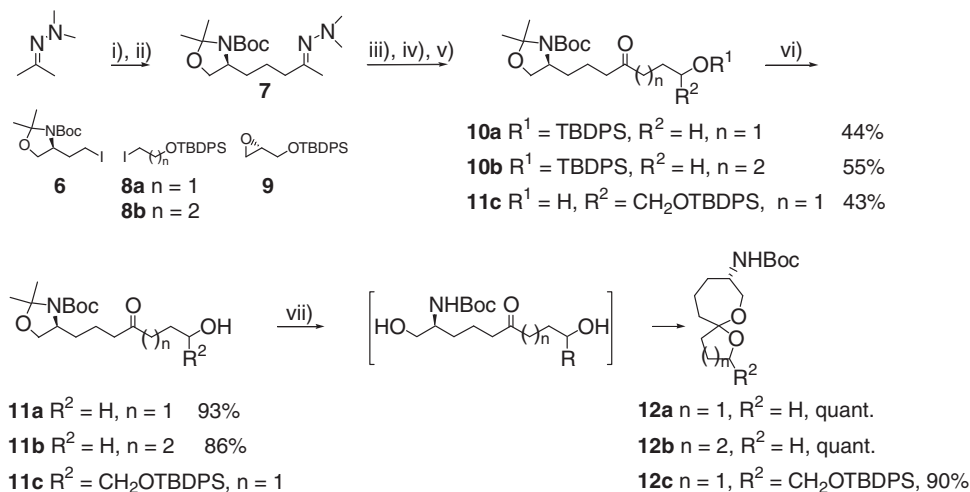
To further evaluate the scope of this sequence, we decided to extend it to the preparation of 1,6-dioxaspiro[4.6]undecane and 1,7-dioxaspiro[5.6]dodecane systems. Herein are disclosed details of our investigations in this area. In this course, we chose to prepare the spiro precursors **11** (Scheme 2).



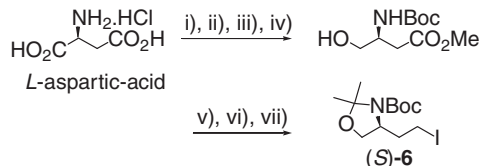
Scheme 1. Synthesis of 1,7-dioxaspiro[5.5]undecanes and 1,6-dioxaspiro[4.5]decanes.

* Corresponding authors. Tel.: +33 473 405 284, +33 473 407 875; fax: +33 473 407 717.

E-mail addresses: M-Eve.Sinibaldi-troin@univ-bpclermont.fr (M.-E. Sinibaldi), isabelle.canet@univ-bpclermont.fr (I. Canet).



Scheme 2. (i) *n*-BuLi, THF, -10 °C; (ii) **6**, THF, quant.; (iii) LDA, -78 °C, THF; (iv) **8a**, **8b** or **9**, -20 °C, 12 h; (v) SiO₂/H₂O, CH₂Cl₂, 20 °C, 48 h, (vi) for **10a,b**, TBAF, THF, 20 °C, quant.; (vii) Yb(OTf)₃, CH₃CN, 20 °C, 24 h.



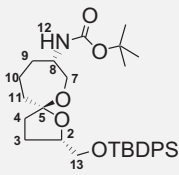
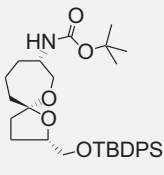
Scheme 3. Synthesis of iodide (*S*)-**6**. (i) SOCl₂, -10 °C, MeOH, 91%; (ii) Boc₂O, Na₂CO₃, dioxane/H₂O, 20 °C, 12 h, 89%; (iii) *N*-hydroxysuccinimide, DCC, CH₂Cl₂, 20 °C, 12 h; (iv) NaBH₄, THF/EtOH (3/1), 0 °C, 15 min, 89% in two steps; (v) DMPE, *p*-TsOH, CH₂Cl₂, 20 °C, 87%; (vi) LiAlH₄, THF, 0 °C then 20 °C, 12 h, quant.; (vii) I₂, imidazole, PPh₃, Et₂O/CH₃CN (3/2), 20 °C, 48 h, 95%.

Thus, treatment of the lithiated acetone *N,N*-dimethylhydrazone by iodide **6** (see Scheme 3 for an access to **6** by an adaptation of the reported synthesis⁷ of this substrate) led quantitatively (¹H NMR of the crude) to hydrazone intermediate **7**. Compound **7** submitted to basic treatment (LDA) underwent a second alkylation step with the known iodides **8a**,⁸ furnishing the disubstituted hydrazones that delivered after their selective SiO₂/H₂O induced cleavage, ketones **10a** and **10b** in 44% and 55% yields, respectively (three steps, one final purification). Subsequent removal of the silyl ether protecting group of **10a,b** was then classically accomplished using TBAF in THF and afforded the key hydroxy-ketones **11a** and

Table 1
¹³C and ¹H NMR data for **12a** and **12b** isomers (CDCl₃)

		(5R,8S)-12a		(5S,8S)-12a		(6R,9S)-12b		(6S,9S)-12b
	δ_{H}, J in Hz	δ_{C}	δ_{H}, J in Hz	δ_{C}	δ_{H}, J in Hz	δ_{C}	δ_{H}, J in Hz	δ_{C}
2	3.63, td (8.0, 5.0) 3.66, br t (7.5)	66.8	3.67, td (8.0, 6.0) 3.77, td (8.0, 6.5)	66.9	3.42, dd (11.0, 5.0) 3.53, td (12.0, 2.5)	61.0	3.49, dd (10.0, 4.0) 3.66, m	61.1
3	1.41, m 1.71, m	24.9	1.46, m 1.80, m	24.7	1.14, m 1.33, qt (13.0, 4.0)	25.7	1.20, m 1.39, m	25.8
4	1.22, m 1.66, m	38.1	1.29, m 1.89, m	38.2	1.21, m 1.64, qt (13.5, 3.5)	19.2	1.33, m 1.85, m	19.2
5		110.5		110.1	1.02, td (13.0, 4.0) 1.44, m	35.2	1.12, m 1.71, m	35.4
6						100.0		99.6
7	3.37, dt (12.5, 2.5) 3.68, dd (13.0, 1.0)	64.0	3.45, dd (12.0, 10.0) 3.60, br d (12.0)	65.4				
8	3.85, dq (8.0, 2.0)	49.7	3.91, m	51.1	3.32, dt (12.5, 2.5) 3.60, d (12.5)	62.9	3.37, td (11.0, 5.0) 3.62, m	64.2
9	1.11, m 1.93, br d (13.5)	35.7	0.90, dt (11.0, 10.5) 1.65, m	36.1	3.88, m	49.7	3.87, m	51.0
10	1.14, m 1.23, m	19.2	1.21, m 1.37, m	20.5	1.10, m 1.92, br d (13.0)	35.9	0.74, m 1.68, m	36.5
11	1.63, dt (14.5, 7.5) 1.77, dd (14.5, 11.0)	38.6	1.65, m 1.85, m	38.1	1.10, m 1.21, m	18.1	1.22, m	19.6
12	5.38, d (8.0)		4.20, br s		1.53, d (11.0) 1.59, m	41.7	1.68, m	41.2
13		155.1		155.0	5.41, d (8.5)		4.07, br s	
14		78.6		78.5		155.1		155.0
15	1.47, s	28.6	1.43, s	28.5		78.6		78.6
16					1.47, s	28.6	1.45, s	28.5

Table 2
 ^{13}C and ^1H NMR data for **12c** isomers

		(2S,5S,8S)- 12c		(2S,5R,8S)- 12c
	δ_{H}, J in Hz	δ_{C}	δ_{H}, J in Hz	δ_{C}
2	4.18, dq (8.0, 4.5)	78.9	4.21, q (6.5)	80.5
3	1.68, m	26.9	1.67, m	27.8
4	2.01, ddt (14.5, 11.5, 8.5)			
	1.51, dt (11.0, 8.0)	38.0	1.16, m	38.2
	1.91, m		1.69, m	
5		110.8		110.5
7	3.60, m	65.5	3.11, br d (12.5)	63.7
	3.47, dd (11.5, 10.0)		3.53, d (13.0)	
8	3.91, m	51.31	3.81, m	49.6
9	1.66, m	36.2	1.08, m	35.6
	0.95, m		1.91, br d (13.5)	
10	1.23, m	20.4	1.19, m	19.2
	1.40, m			
11	1.70, m	38.3	1.60, dd (15.0, 7.5)	38.5
	1.91, m		1.76, dd (15.0, 11.5)	
12	4.26, br d (8.0)		5.29, d (8.5)	
13	3.68, dd (10.5, 4.0)	66.7	3.53, dd (10.0, 5.5)	68.7
	3.58, dd (10.5, 4.5)		3.78, dd (10.0, 6.5)	
N-Boc	1.43, s	155.0, 78.6, 28.5	1.46, s	155.0, 78.5, 28.6
Si-tBu	1.15, s	19.6, 27.1	1.17, s	

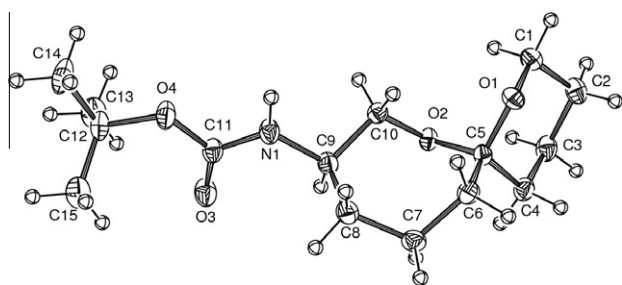


Figure 1. ORTEP of **12b**.

11b in good yields. In a same manner, the attack of the anion of hydrazone **7** on epoxide **9** permitted its ring opening, leading directly to the keto-diol **11c** in 43% yield.

Finally, *p*-TsOH/MeOH mediated deprotection-spirocyclization of **11a,b,c** afforded, as the sole cleanly isolated derivatives, 1,6-dioxaspiro[4.6]undecanes **12a,c** and 1,7-dioxaspiro[5.6]dodecane **12b** (Scheme 2). In an attempt to increase the formation of these spirocyclic compounds, we modified the nature of the acid. Treatment of **11** by HCl/methanol or Amberlyst[®] 15 led unfortunately to degradation of the reaction mixture. Conversely, the use of Yb(OTf)₃ in CH₃CN revealed efficient^{3d} and gave nearly quantitatively and exclusively the spirocyclic epimers **12** in a 1/1 ratio.

These diastereomers could be cleanly separated on neutral alumina and fully characterized (Tables 1 and 2). Furthermore, in the spiro[5.6] series, if the (6R,9S)-isomer of **12b** occurred as an oil, its (6S,9S) epimer could be obtained as a fine white powder easily recrystallized from ethanol. Its structure and the relative stereochemistry of the spiranic center were then confirmed by an X-ray crystallographic analysis as shown as ORTEP plot in Figure 1.¹⁰

In summary, we demonstrated here the efficiency of our strategy to prepare in a few steps and good yields the 1,6-dioxaspiro[4.6]undecanes **12a,c** (five steps, 41% yield for **12a**; four steps, 39% yield for **12c**) and 1,7-dioxaspiro[5.6]dodecane **12c** (five steps,

47% yield) as a 1/1 separable mixture of the two spiranic epimers. Our approach allowed the introduction of substitutions on the β and γ positions relative to the spiranic center, using simple iodides or epoxide precursors. Extension to the synthesis of others spiroketals will be reported in due course.

References and notes

- (a) Perron, F.; Albizati, K. F. *Chem. Rev.* **1989**, *89*, 1617; (b) Jacobs, M. F.; Kitching, W. *Curr. Org. Chem.* **1998**, *2*, 395; (c) Francke, W.; Schröder, W. *Curr. Org. Chem.* **1999**, *3*, 407; (d) Francke, W.; Kitching, W. *Curr. Org. Chem.* **2001**, *5*, 233.
- (a) Faul, M. M.; Huff, B. E. *Chem. Rev.* **2000**, *100*, 2407; (b) Mead, K. T.; Brewer, B. N. *Curr. Org. Chem.* **2003**, *7*, 227; (c) Aho, J. E.; Pihko, P. M.; Rissa, T. K. *Chem. Rev.* **2005**, *105*, 4406.
- For recent examples of synthesis of these systems, see: (a) Aponick, A.; Li, C. Y.; Palmes, J. A. *Org. Lett.* **2009**, *11*, 121; (b) Selvaratnam, S.; Ho, J. H. H.; Huelatt, P. B.; Messerle, B. A.; Chai, C. L. L. *Tetrahedron Lett.* **2009**, *50*, 1125; (c) Liu, G.; Wurst, J. M.; Tan, D. S. *Org. Lett.* **2009**, *11*, 3670; (d) Aponick, A.; Li, C.-Y.; Palmes, J. A. *Org. Lett.* **2009**, *11*, 121; (e) de Greef, M.; Zard, S. Z. *Org. Lett.* **2007**, *9*, 1773; (f) Moilanen, S. B.; Potuzak, J. S.; Tan, D. S. *J. Am. Chem. Soc.* **2006**, *128*, 1792; (g) Doubsky, J.; Streinz, L.; Saman, D.; Zednik, J.; Koutek, B. *Org. Lett.* **2004**, *6*, 4909; (h) Ghosh, S. K.; Hsung, R. P.; Wang, J. *Tetrahedron Lett.* **2004**, *45*, 5505; (i) Bez, G.; Bezbarua, M. S.; Saikia, A. K.; Barua, N. C. *Synthesis* **2000**, 537; (j) Nakamura, K.; Kitayama, T.; Inoue, Y.; Ohno, A. *Tetrahedron* **1990**, *46*, 7471.
- For a review of the use of *N,N*-dialkylhydrazones in organic synthesis see: (a) Lazny, R.; Nodzevska, A. *Chem. Rev.* **2010**, *110*, 1366; For some recent applications in spiroketals synthesis see: (b) Dias, L. C.; de Oliveira, L. G. *Org. Lett.* **2004**, *6*, 2587; (c) Panek, J. S.; Jain, N. F. *J. Org. Chem.* **2001**, *66*, 2747.
- Tursun, A.; Canet, I.; Aboab, B.; Sinibaldi, M.-E. *Tetrahedron Lett.* **2005**, *46*, 2291.
- Goubert, M.; Canet, I.; Sinibaldi, M.-E. *Eur. J. Org. Chem.* **2006**, 4805.
- Kadota, I.; Saya, S.; Yamamoto, Y. *Heterocycles* **1997**, *46*, 335.
- Compounds **7a,b** were prepared from their monosilylated alcohol (Mac Dougal, P. G.; Rico, J. G.; Oh, Y.-I.; Condon, B. D. *J. Org. Chem.* **1986**, *51*, 3388). Spectroscopic data were in accordance with those reported by Paquette, L. A.; Doherty, A. M.; Rayner, C. M. *J. Am. Chem. Soc.* **1992**, *114*, 3910 for **8a**, by Dawson, I. M.; Gregory, J. A.; Herbert, R. B.; Sammes, P. G. *J. Chem. Soc. Perkin Trans. 1* **1988**, 2585 for **8b**.
- (a) Nicolaou, K. C.; He, Y.; Vourloumis, D.; Valberg, H.; Roschangar, F.; Sarabia, F.; Ninkovic, S.; Yang, Z.; Trujilla, J. I. *J. Am. Chem. Soc.* **1997**, *34*, 7960; For ring opening of epoxide by *N,N*-dimethylhydrazones see: (b) Evans, D. A.; Bender, S. L.; Morris, J. J. *J. Am. Chem. Soc.* **1988**, *110*, 2506.
- The crystal structure has been deposited at the Cambridge Crystallographic Data Centre and allocated the deposition number CCDC 720349.